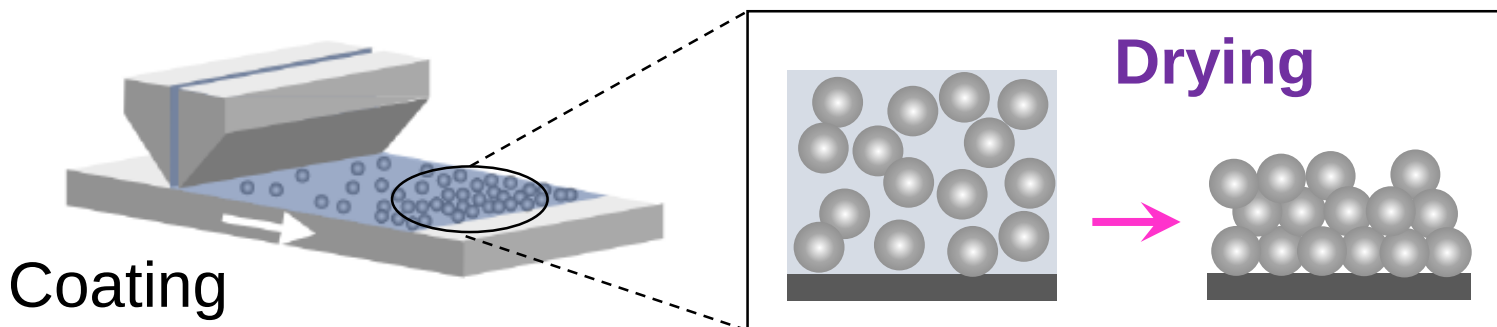


# Structure control based on the drying characteristics of colloidal suspensions

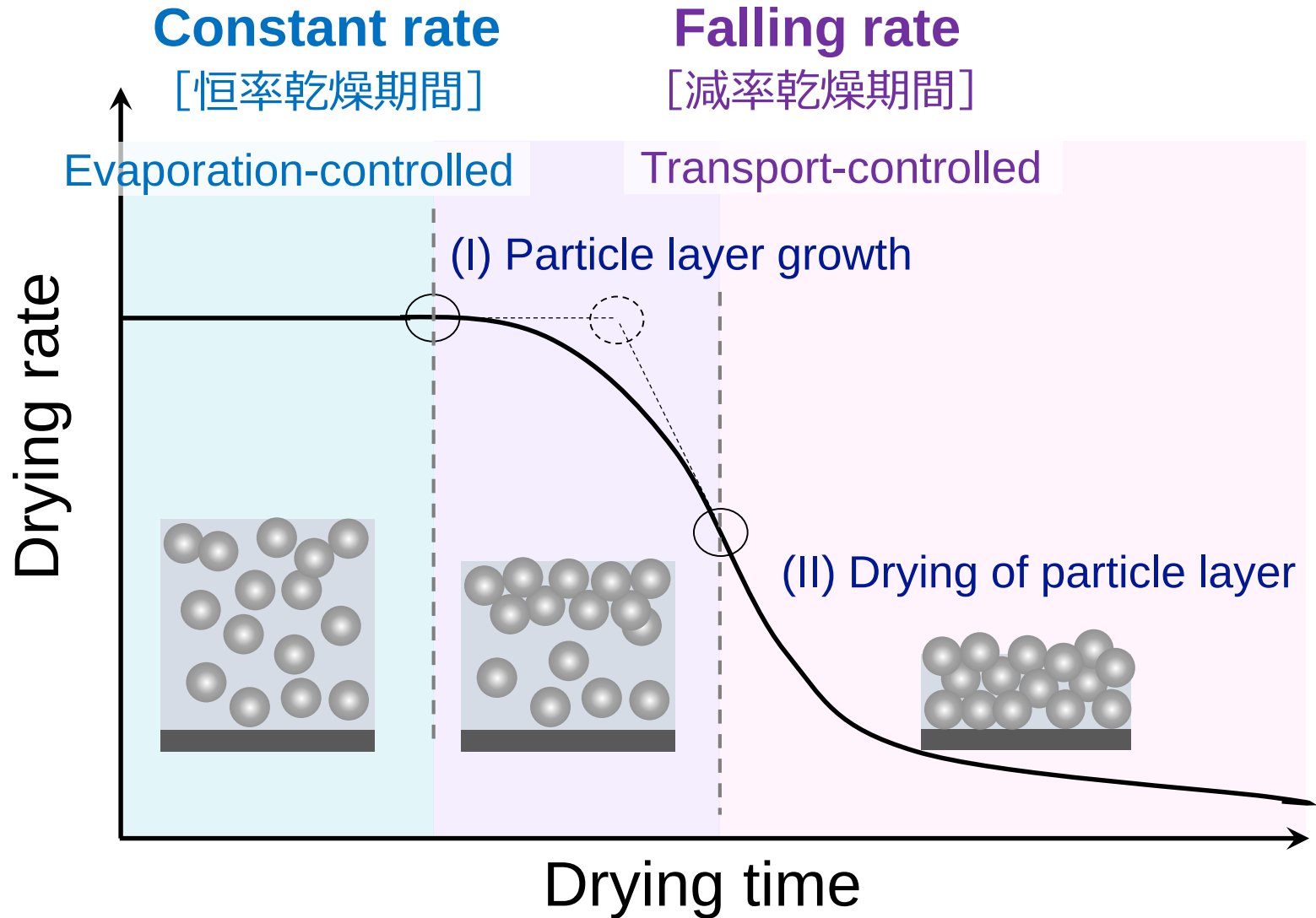
## 微粒子分散液の乾燥特性に基づく構造制御の提案

- 辰巳 怜 (東大環安セ)
- 小池 修 (PIA)
- 山口由岐夫 (PIA)
- 辻 佳子 (東大環安セ/東大院工)



# Drying Curve of Colloidal Suspensions

Drying characteristics  $\leftrightarrow$  Structure



# Skinning

## Heat balance

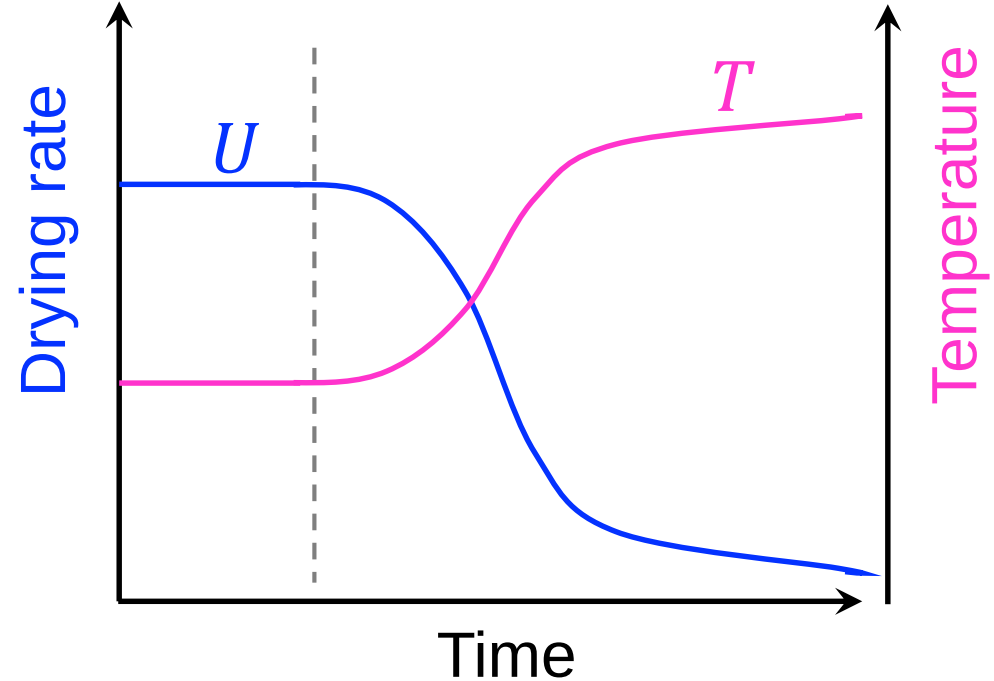
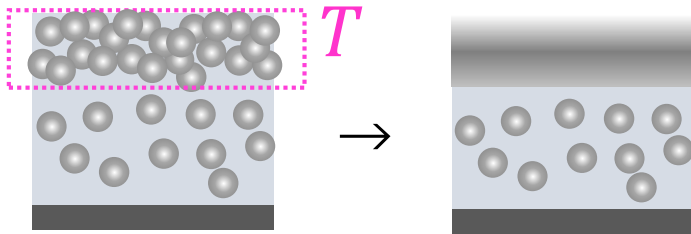
$$\rho U \Delta H_v = h(T - T_{\text{ex}})$$

Liquid density:  $\rho$

Latent heat:  $\Delta H_v$

Heat transfer coefficient:  $h$

Drying temperature:  $T_{\text{ex}}$



減率乾燥期間で材料温度上昇

→ 表面での乾燥, 析出・熱変性

→ スキン層形成 (Skinning)

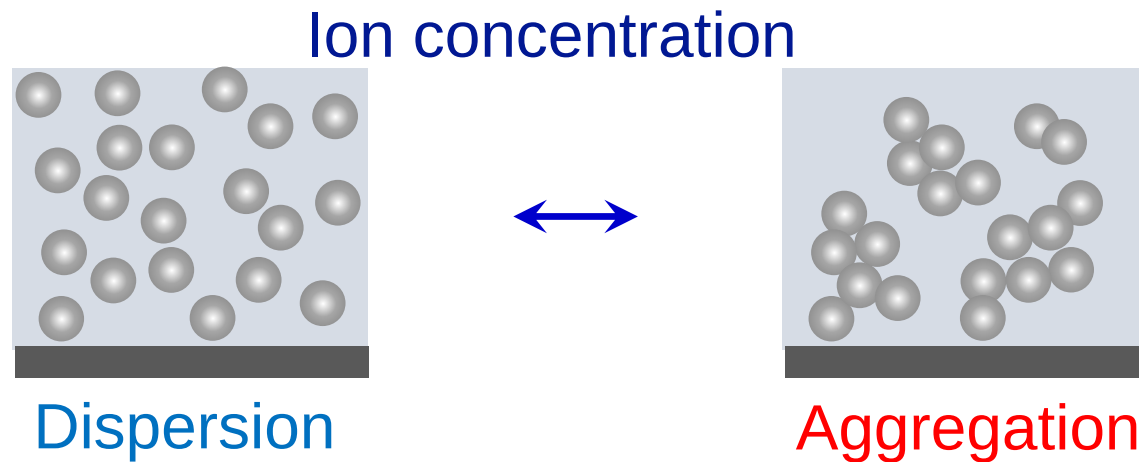
- ・ 透水抵抗 → 乾燥速度の低下
- ・ 材料密度の不均一化
- ・ 表面荒れの誘起

### 対策

- ・ 乾燥温度の制御 (下げる)
- ・ 乾燥速度低下の抑制

# Objective

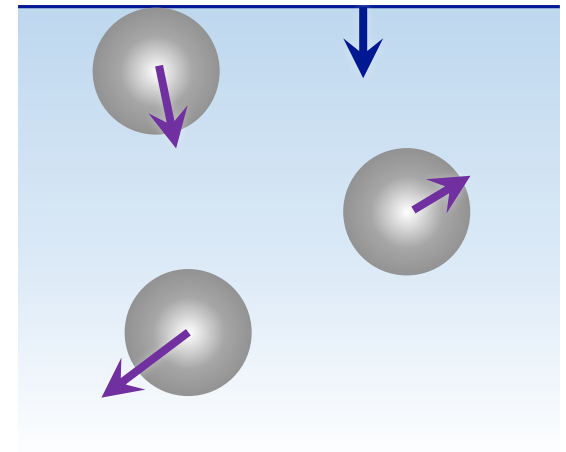
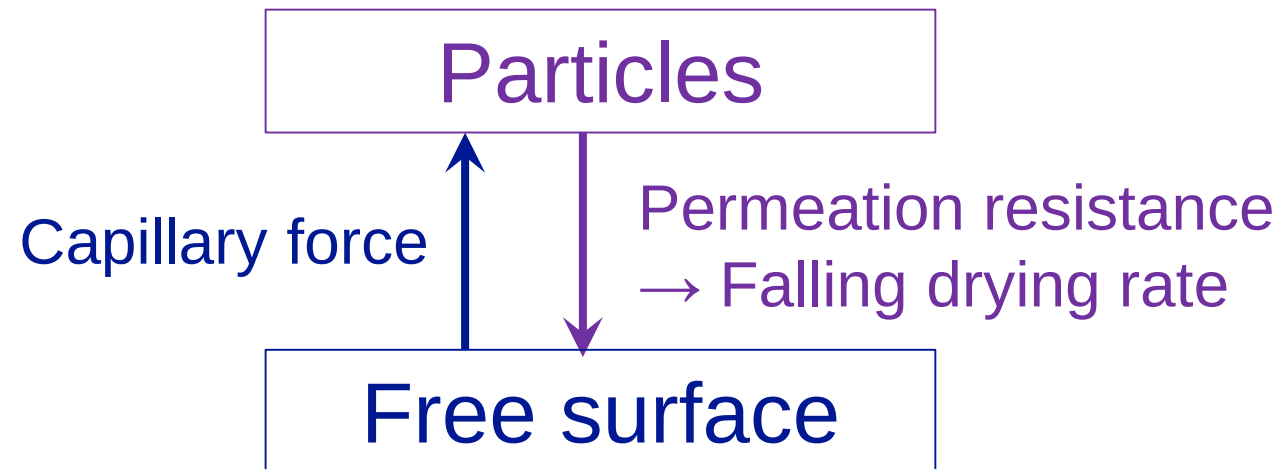
- ◆ Estimation of the drying curve of colloidal suspensions by numerical simulations (SNAP-L)
- ◆ Control of drying characteristics by dispersion/aggregation



Drying characteristics  $\leftrightarrow$  Structure

# Model

- **Particles:** Equation of motion (Langevin eq.)
- **Free surface:** Recession with a varying rate

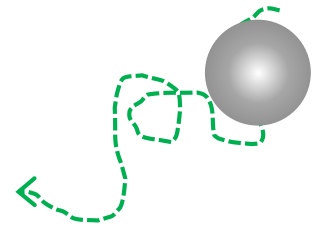


# Particles' Brownian Motion

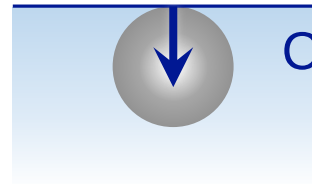
## Langevin equation

$$M\dot{\mathbf{V}} = \underbrace{-\xi\mathbf{V}}_{\text{Fluid}} + \underbrace{\mathbf{F}^R}_{\text{Free surface}} + \underbrace{\mathbf{F}^{\text{cpl}}}_{\text{Free surface}} + \underbrace{\mathbf{F}^{\text{cnt}}}_{\text{Interparticle}} + \mathbf{F}^{\text{DLVO}}$$

- **Drag force:**  $-\xi\mathbf{V}$  (Stokes' law:  $\xi = 3\pi\eta d$ )
- **Random force:**  $F_{\alpha}^R(t) \sim N(0, 2\xi k_B T \Delta t)$  (Gaussian dist.)  
 → **Brownian Diffusion** (Diffusion coefficient:  $D = \frac{k_B T}{3\pi\eta d}$ )

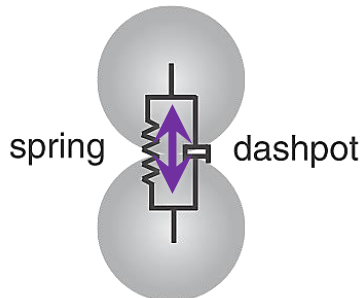


- **Vertical capillary force:**  $F^{\text{cpl}}$

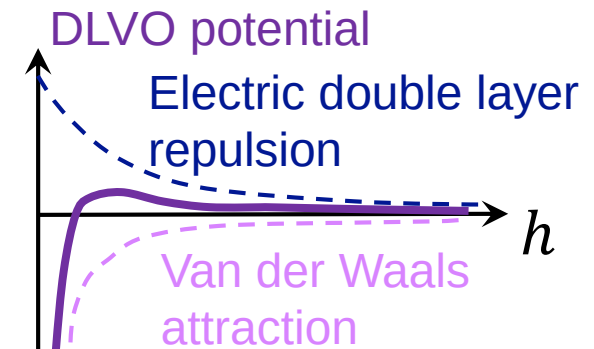
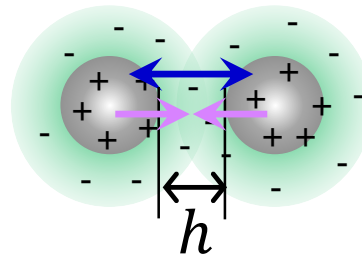


Contact angle  $\alpha = 0$

- **Contact force:**  $F^{\text{cnt}}$



- **DLVO force:**  $F^{\text{DLVO}}$



→ **Dispersion / Aggregation**

# Drying Rate

## Darcy's law

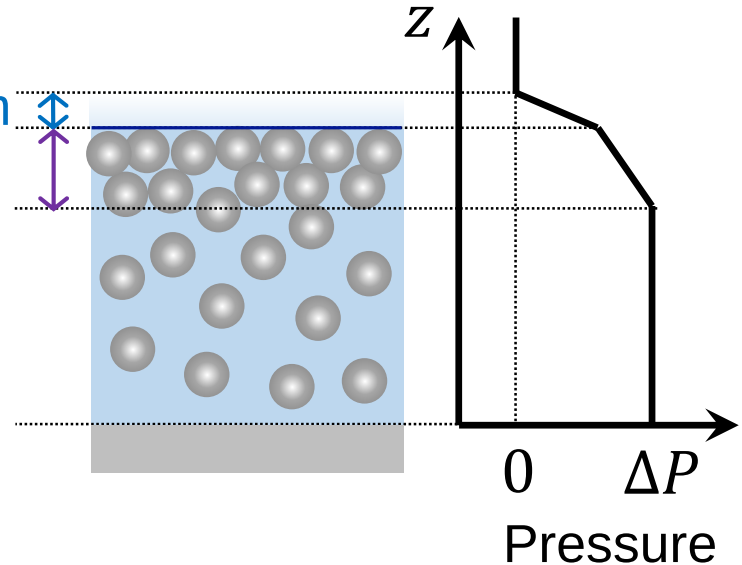
Drying rate:  $U = \frac{\Delta P}{\eta(R_0 + R_p)}$

Resistance of evaporation:  $R_0$

Resistance of particle layer:  $R_p$

Viscosity of liquid:  $\eta$

Boundary film  
Particle layer



$$\rightarrow \frac{U}{U_0} = \frac{R_0}{R_0 + R_p}$$

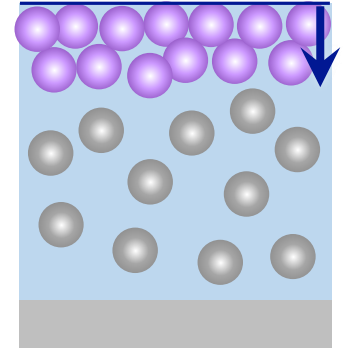
Initial drying rate:  $U_0 = \frac{\Delta P}{\eta R_0}$

(Constant drying period)

# Resistance of Particle Layer

Drying rate:  $\frac{U}{U_0} = \frac{R_0}{R_0 + R_p}$

Particle layer = Aggregation moving with free surface

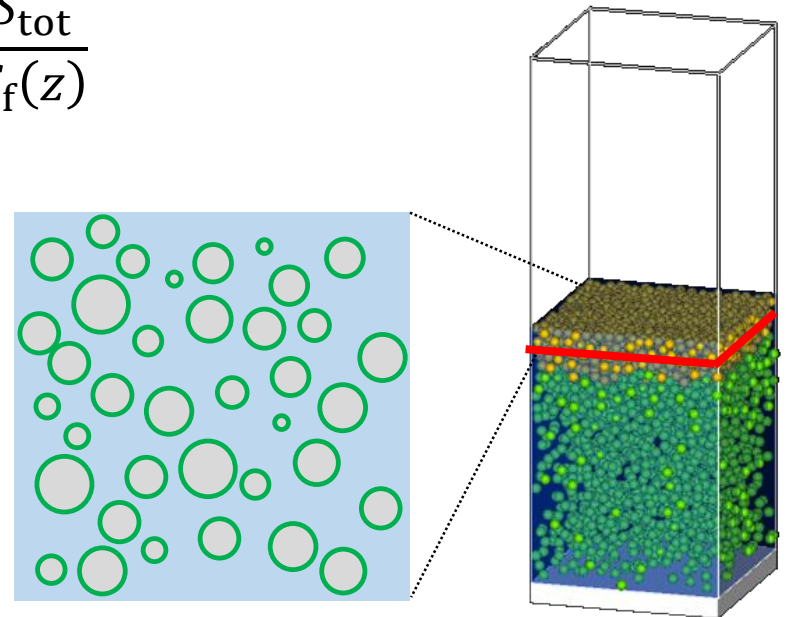


$$R_p = \int_P r(z) dz \quad r(z) = \frac{80}{[D_H(z)]^2} \frac{S_{\text{tot}}}{S_f(z)}$$

Hydraulic diameter:  $D_H = \frac{4S_f}{L_f}$

Cross-sectional area of the flow:  $S_f$

Wetted perimeter:  $L_f$



Cross-section



# Simulation Conditions

## Particles

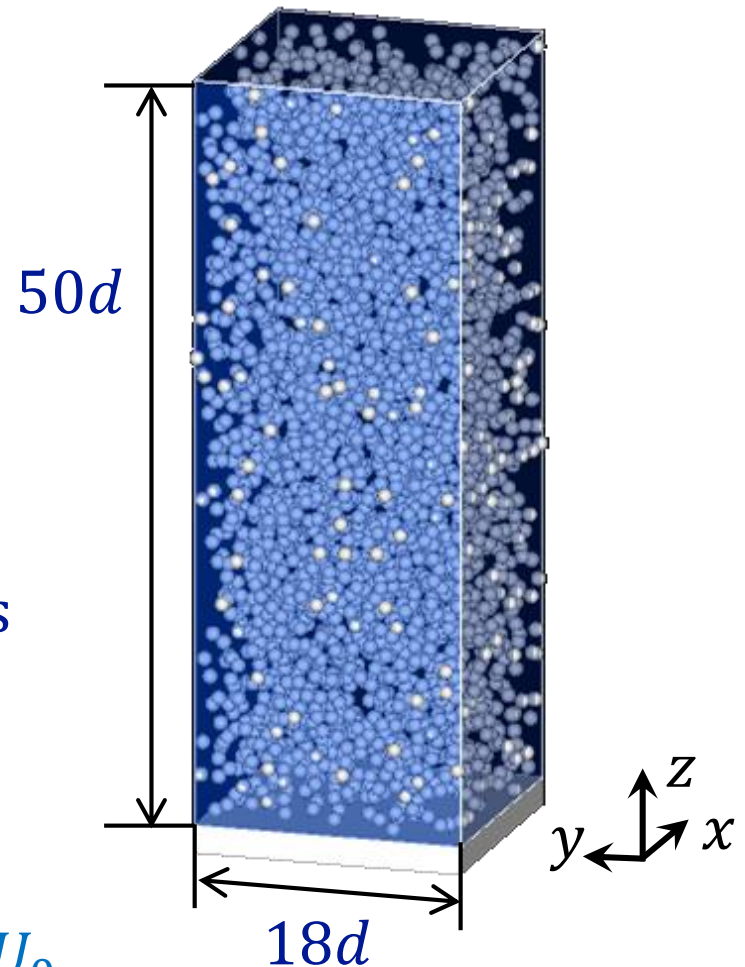
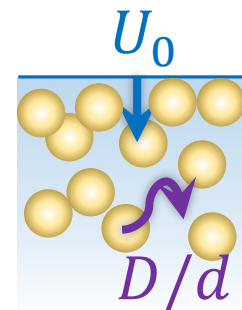
- diameter  $d = 100 \text{ nm}$
- Initial volume particle fraction  $0.1$
- Zeta potential  $30 \text{ mV}$

## Medium

- Water
- Initial drying rate  $U_0 = 5 \times 10^{-3} \text{ m/s}$
- Ion concentration  
 $5 \times 10^{-4} - 1 \times 10^{-2} \text{ M}$

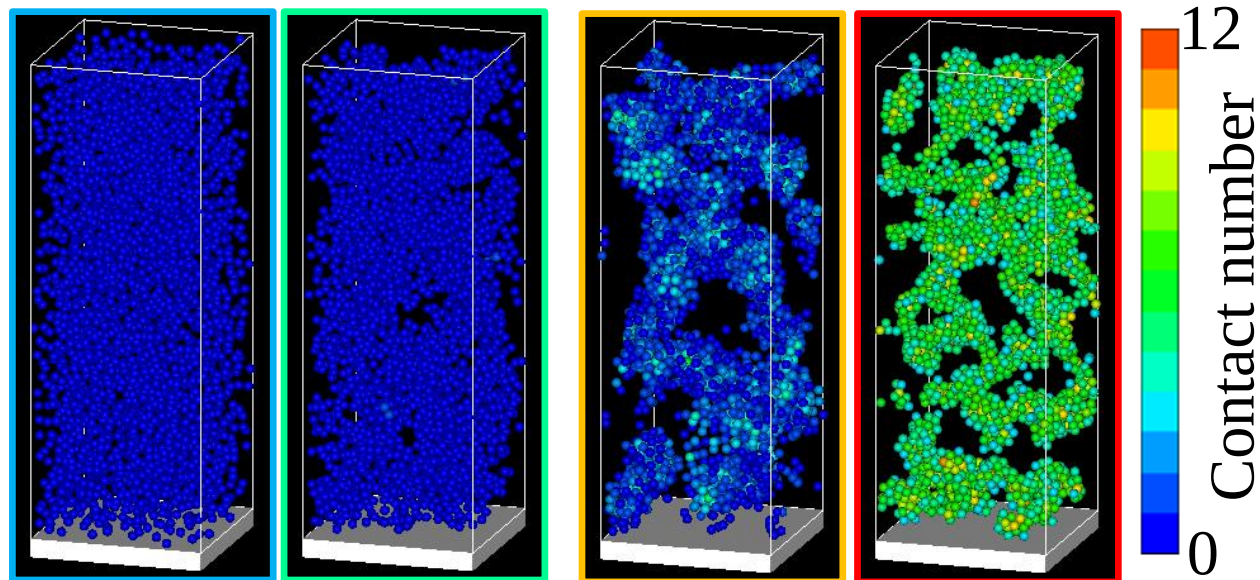
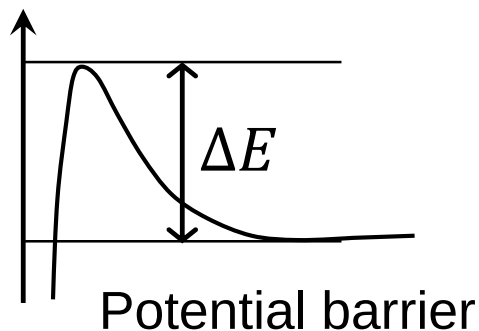
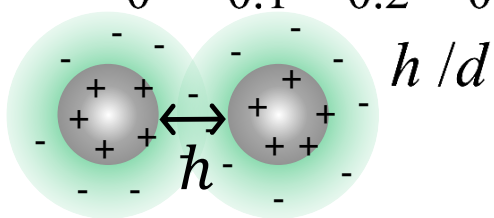
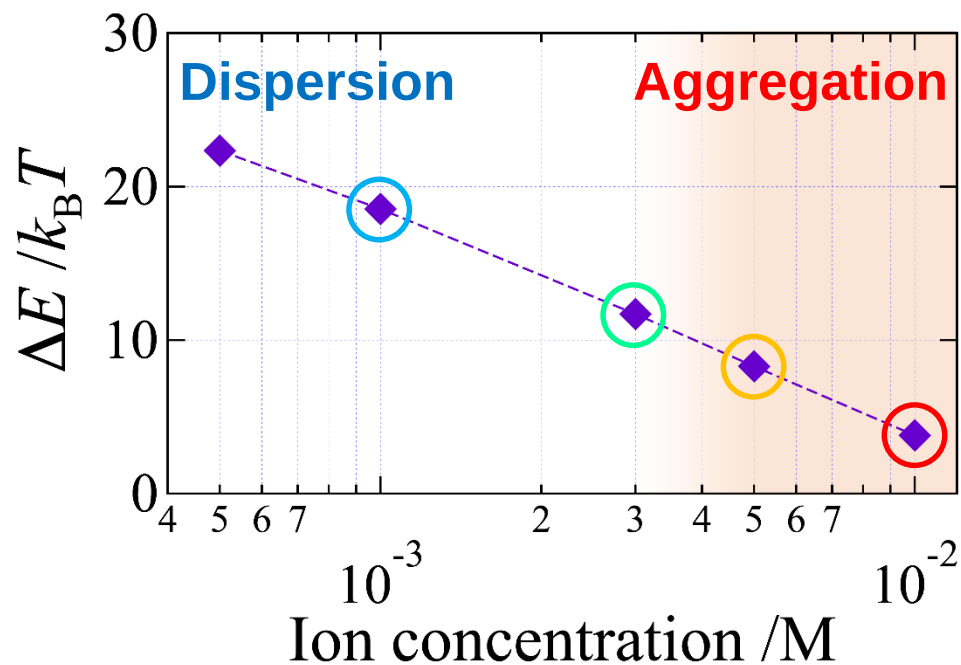
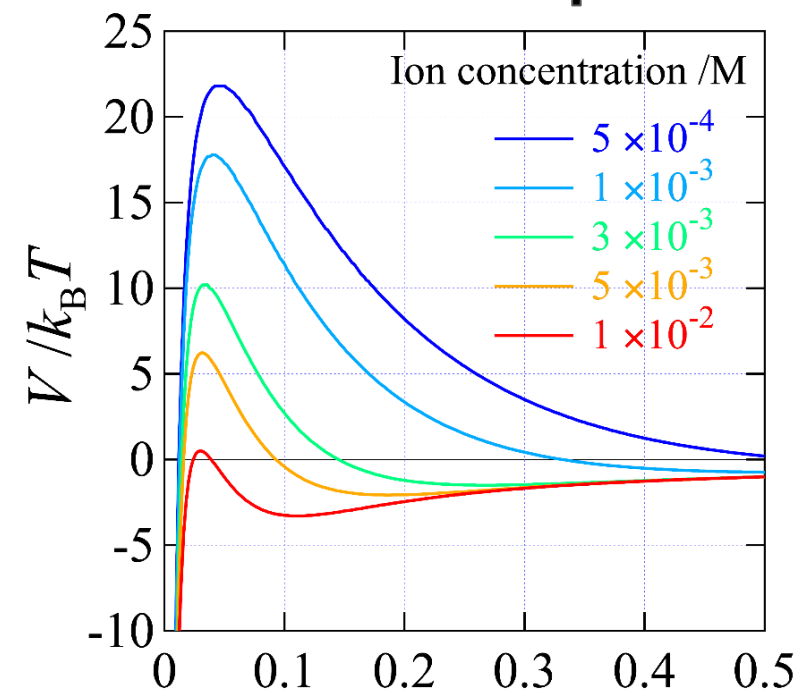
Initial particle drying Péclet number

$$Pe_0 = \frac{(\text{Drying rate})}{(\text{Diffusion rate})} = \frac{U_0}{D/d} = 100$$

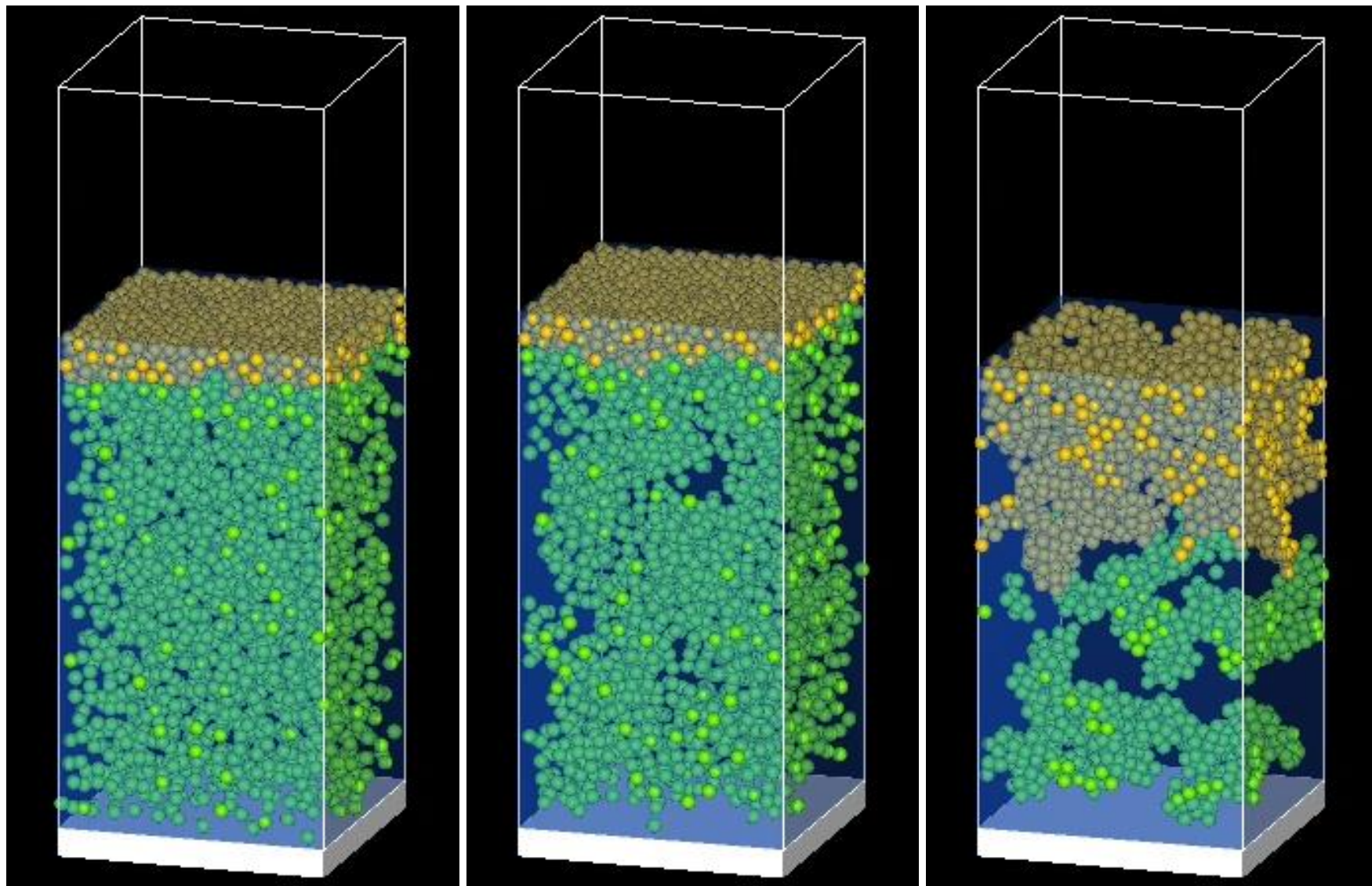


Periodic boundaries  $x, y$

# Interparticle DLVO potential



# Particle Distribution



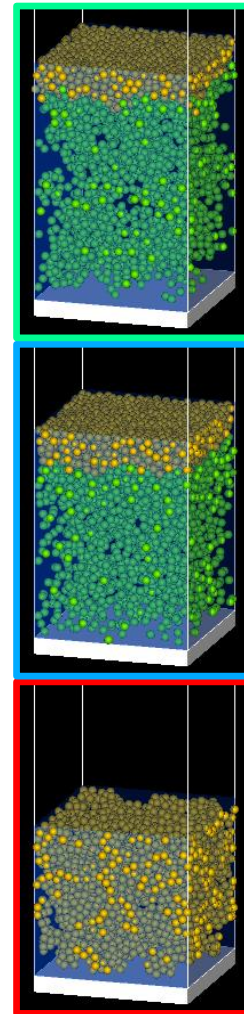
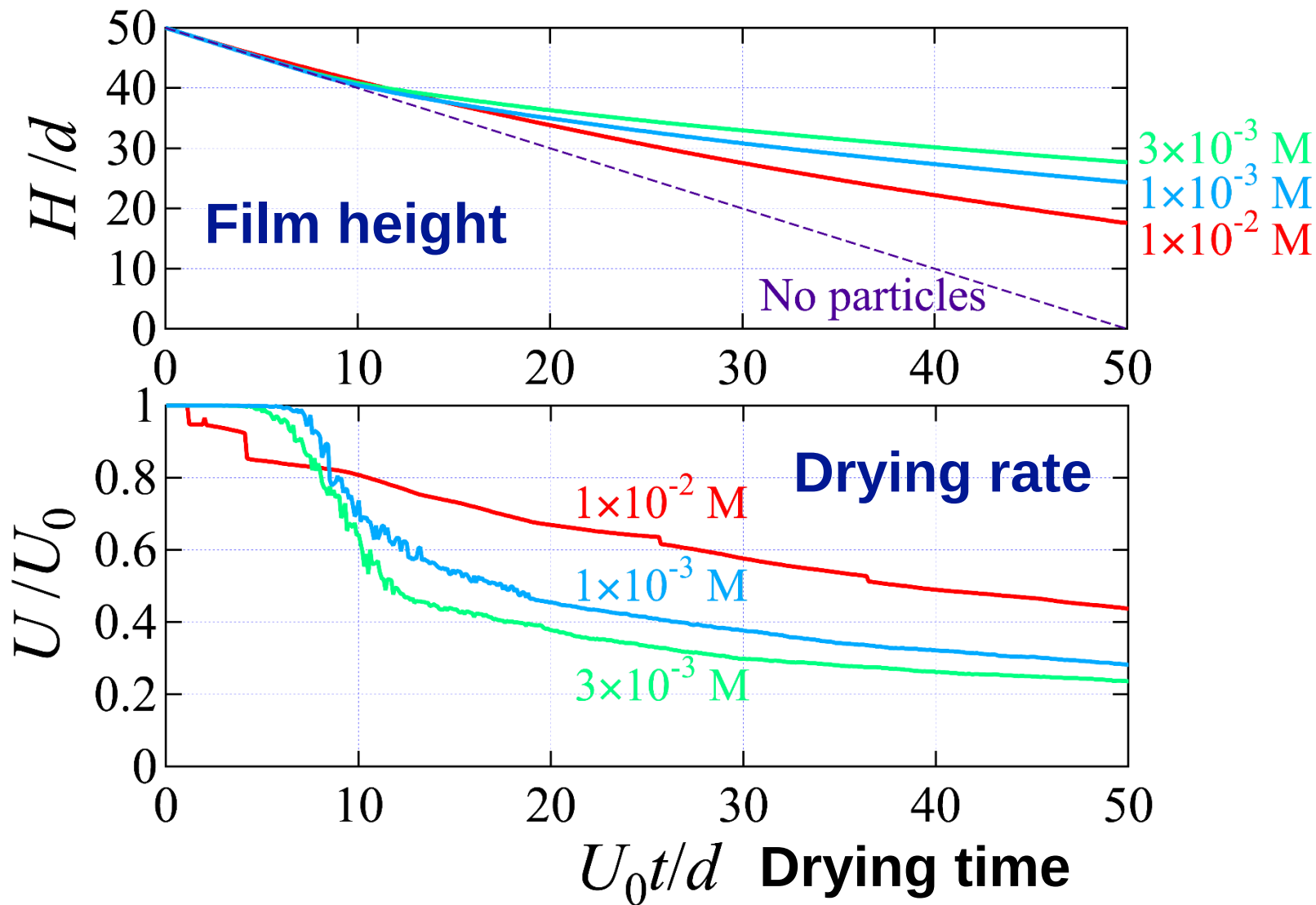
$1 \times 10^{-3} \text{ M}$

$3 \times 10^{-3} \text{ M}$

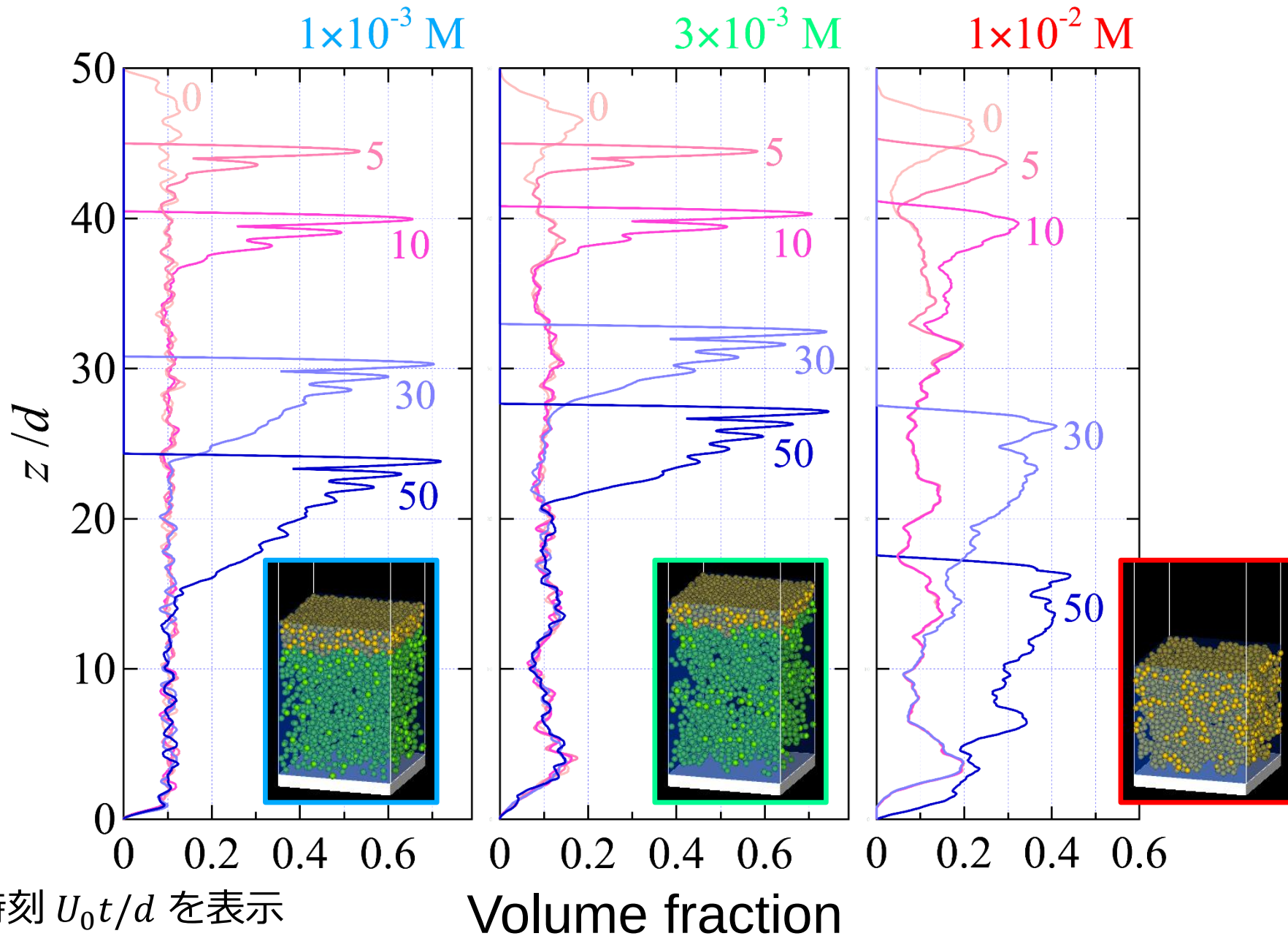
$1 \times 10^{-2} \text{ M}$



# Drying Characteristics



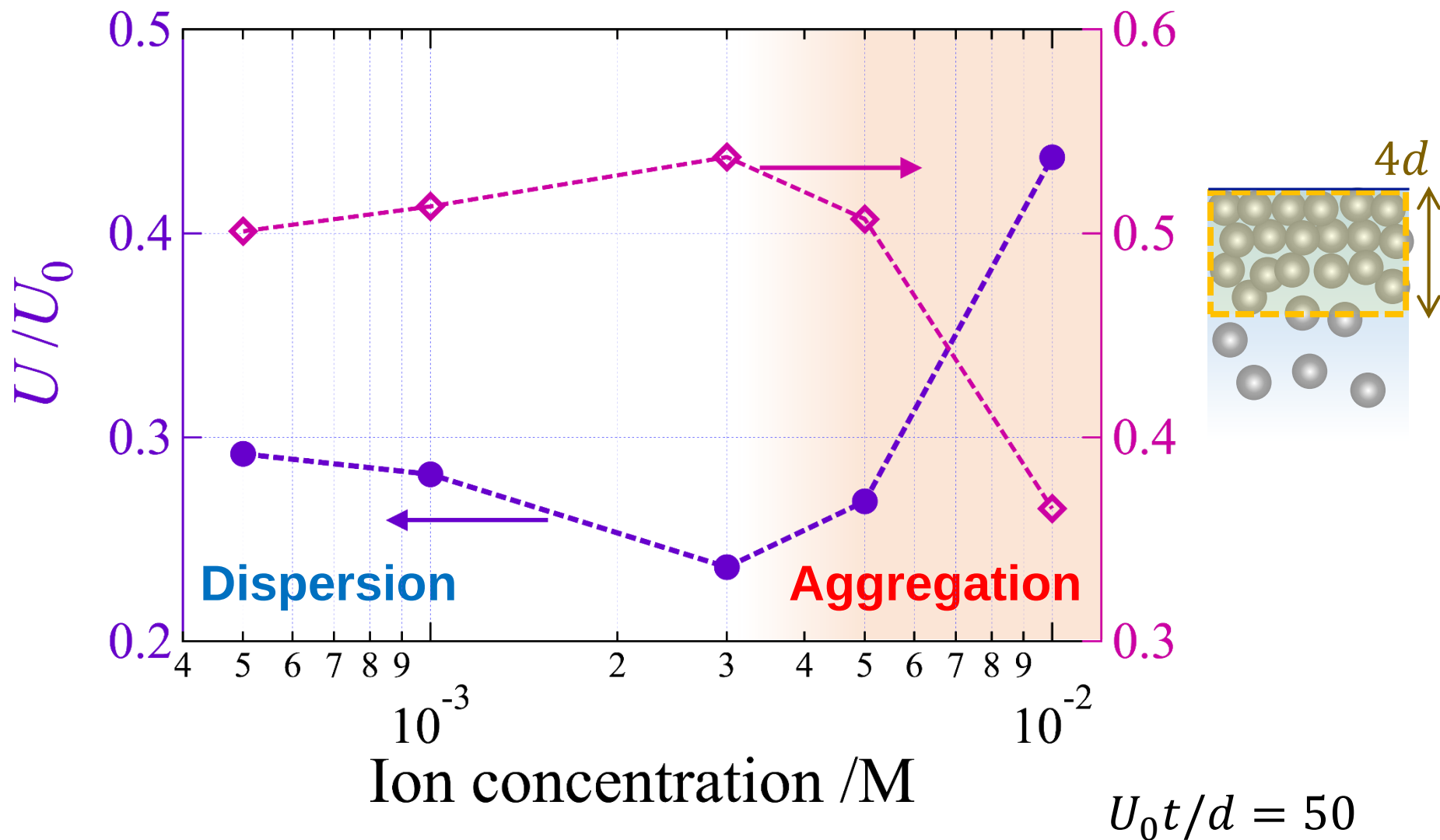
# Particle Distribution



# Summary

Drying rate

Volume fraction of particle layer



# Summary

## ◆ How to prevent skinning

- Drying characteristics → Temperature control
- Dispersion/aggregation control to reduce drying time

Aggregation: higher porosity → higher drying rate

Dispersion: slower particle layer growth  
→ higher drying rate