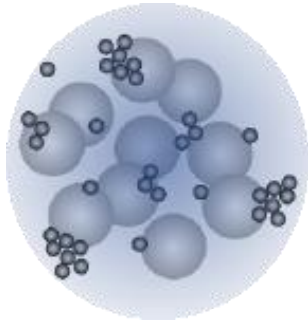


Structure of dried colloidal films controlled by the morphology of aggregated particles

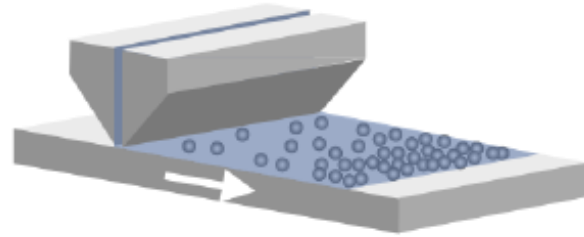
微粒子分散液の凝集状態による乾燥粒子膜の構造制御

- 辰巳 怜 (東大環安セ)
- 小池 修 (PIA)
- 山口由岐夫 (PIA)
- 辻 佳子 (東大環安セ/東大院工)

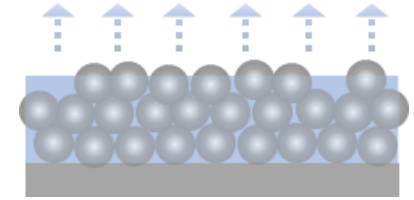
Material Fabrication from Colloidal Suspensions²



Dispersing



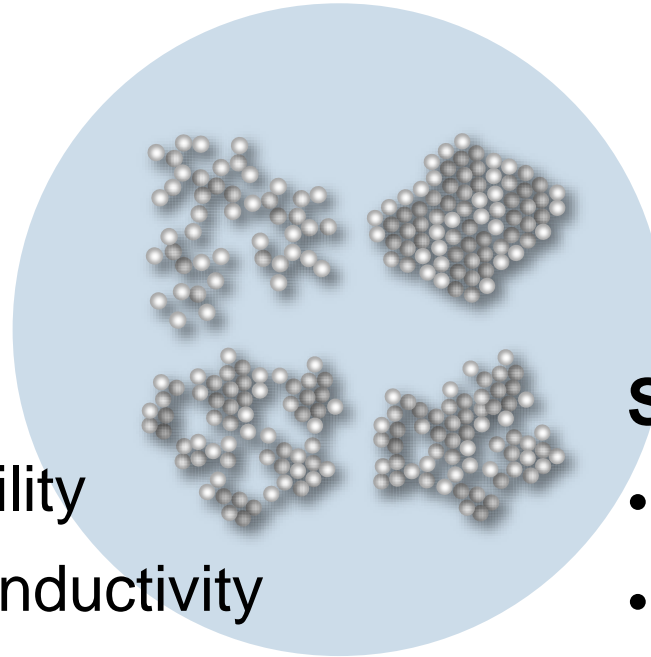
Coating



Drying

Functions

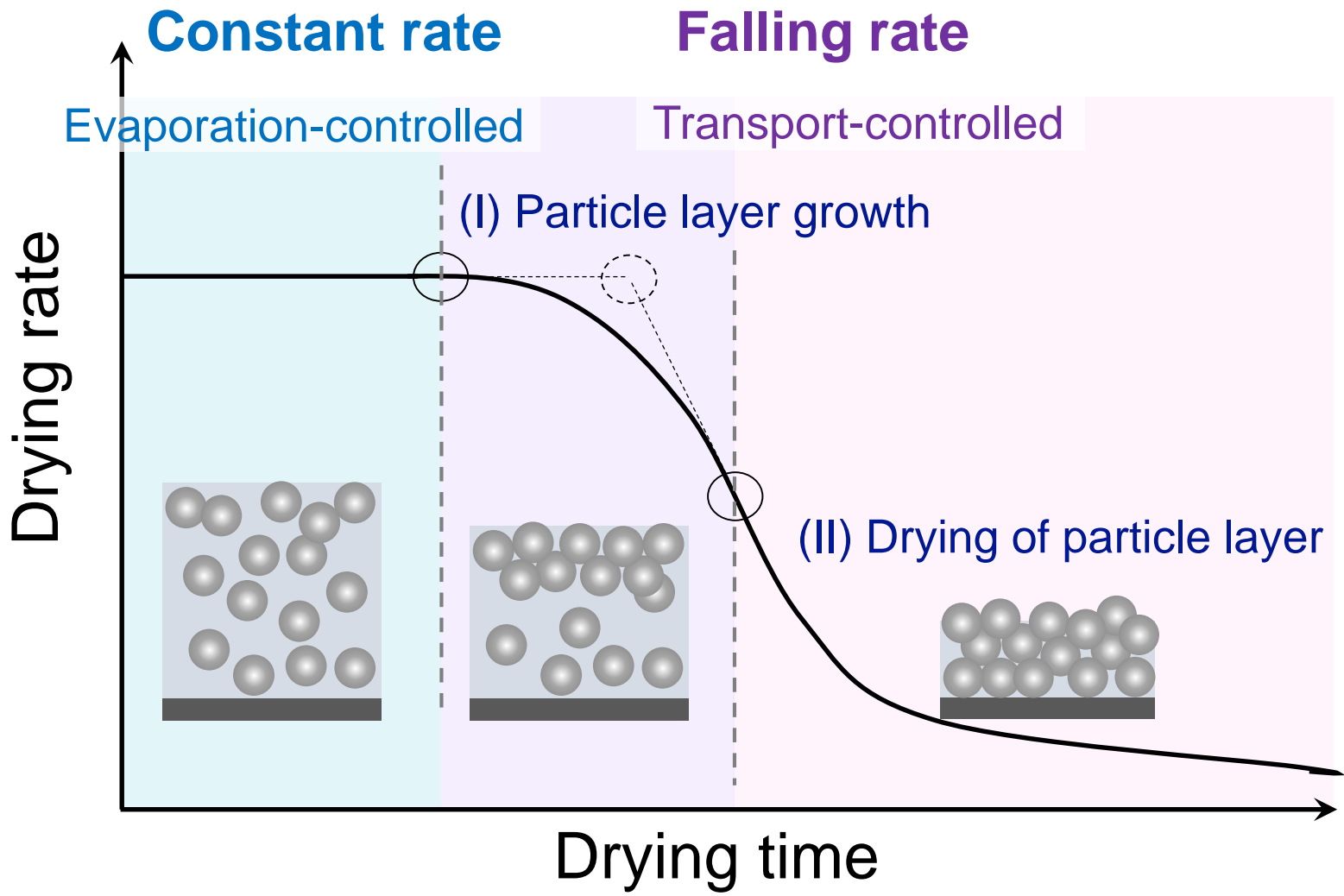
- Strength
- Permeability
- Electrical/Thermal conductivity
- Optical property



Structures

- Porosity
- Contact network

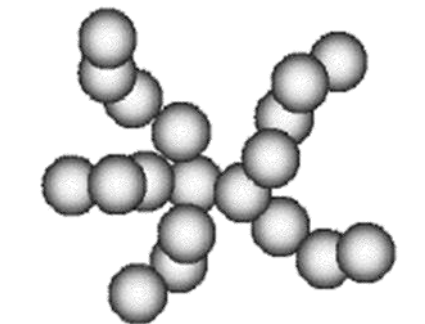
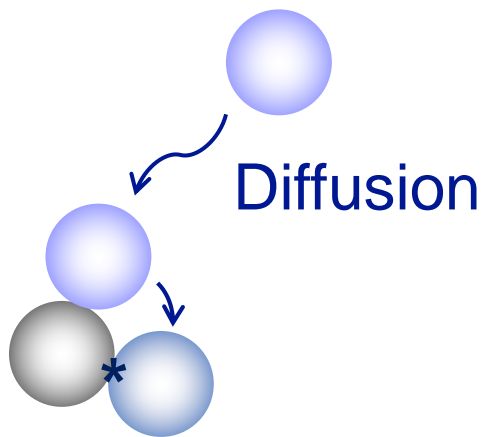
Drying Curve of Colloidal Suspensions



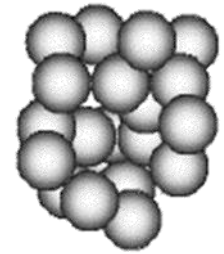
Drying rate vs. Structure

Objective

- ◆ Modeling of adhesion to describe the various morphologies of aggregated particles
- ◆ Effects of adhesion on structure formation during drying



Diffusion-limited



Reaction-limited

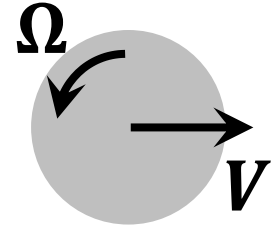
Adhesion: Fixation of contact points
(Reaction)

Equations of Particles' Motion

Langevin equation

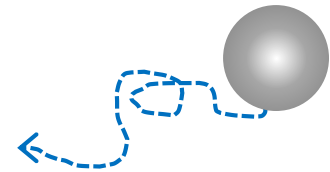
$$M\dot{V} = \underbrace{-\xi V + F^R}_{\text{Hydrodynamic force/torque}} + F^{\text{cnt}} + \underbrace{F^{\text{DLVO}}}_{\text{DLVO force}} + \underbrace{F^{\text{cpl}}}_{\text{Capillary force}}$$

$$I\dot{\Omega} = \underbrace{-\zeta\Omega + N^R}_{\text{Hydrodynamic force/torque}} + N^{\text{cnt}}$$

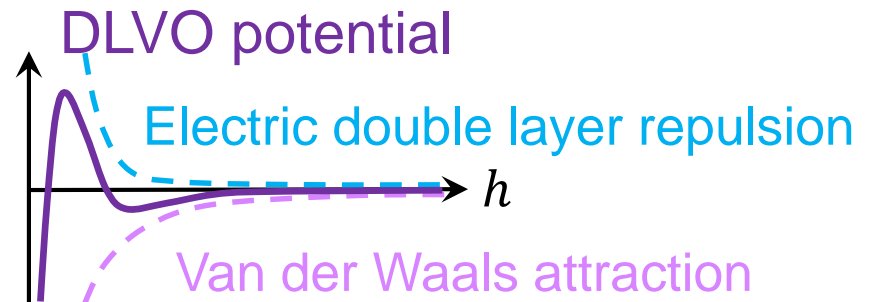
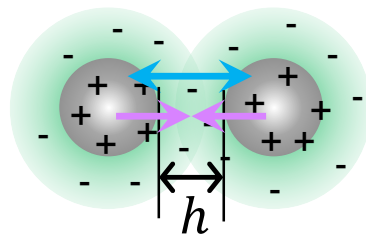


- **Hydrodynamic force/torque**

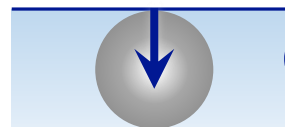
Drag + Fluctuations \rightarrow Brownian motion



- **DLVO force**



- **Capillary force**



Contact angle $\alpha = 0$

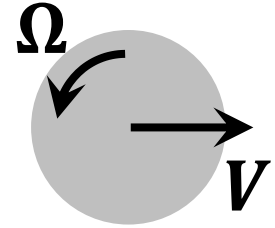
Vertical push into liquid

Modeling of Adhesion

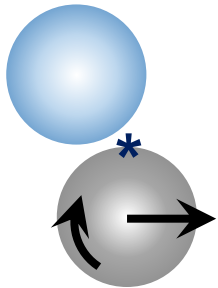
Langevin equation

$$M\dot{V} = -\xi V + F^R + F^{cnt} + F^{DLVO} + F^{cpl}$$

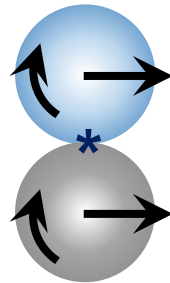
$$I\dot{\Omega} = -\zeta\Omega + N^R + N^{cnt}$$



• Contact force/torque

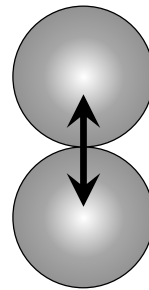


Slip

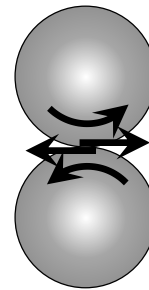


Stick

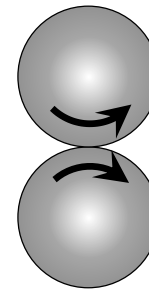
Adhesion



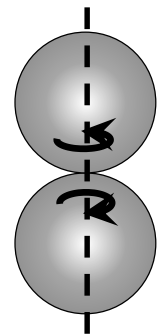
Normal



Shear



Bending



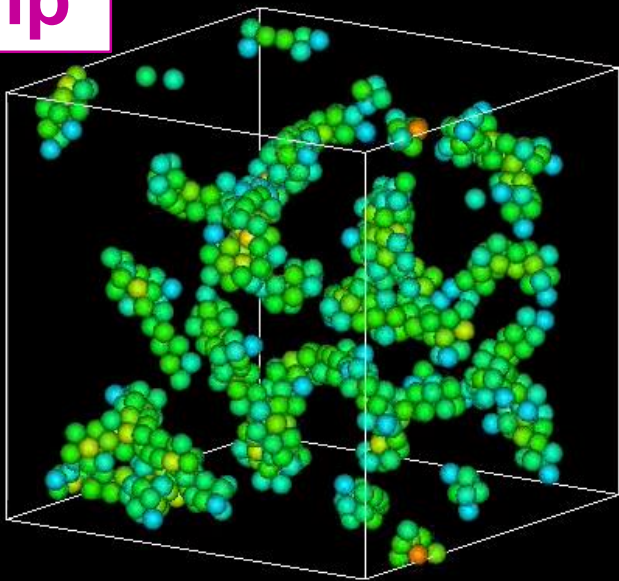
Twist

Constraint on the relative motions

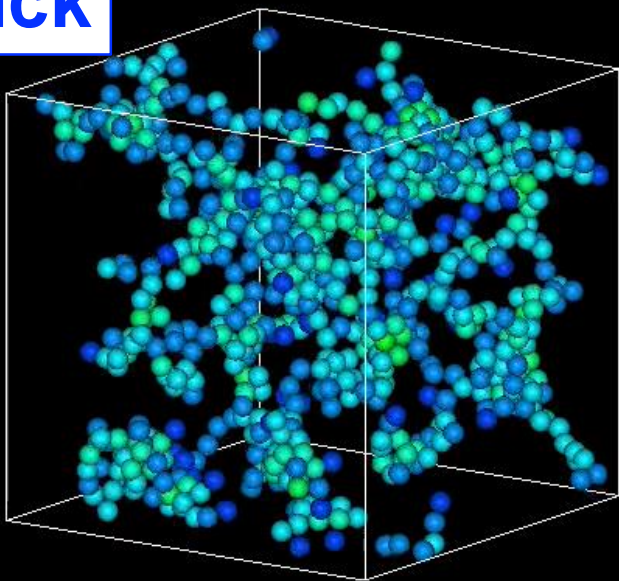
Aggregation

Particle diameter: $d = 10$ nm
 Zeta potential: 0 mV
 Volume fraction: 5 vol%

Slip



Stick

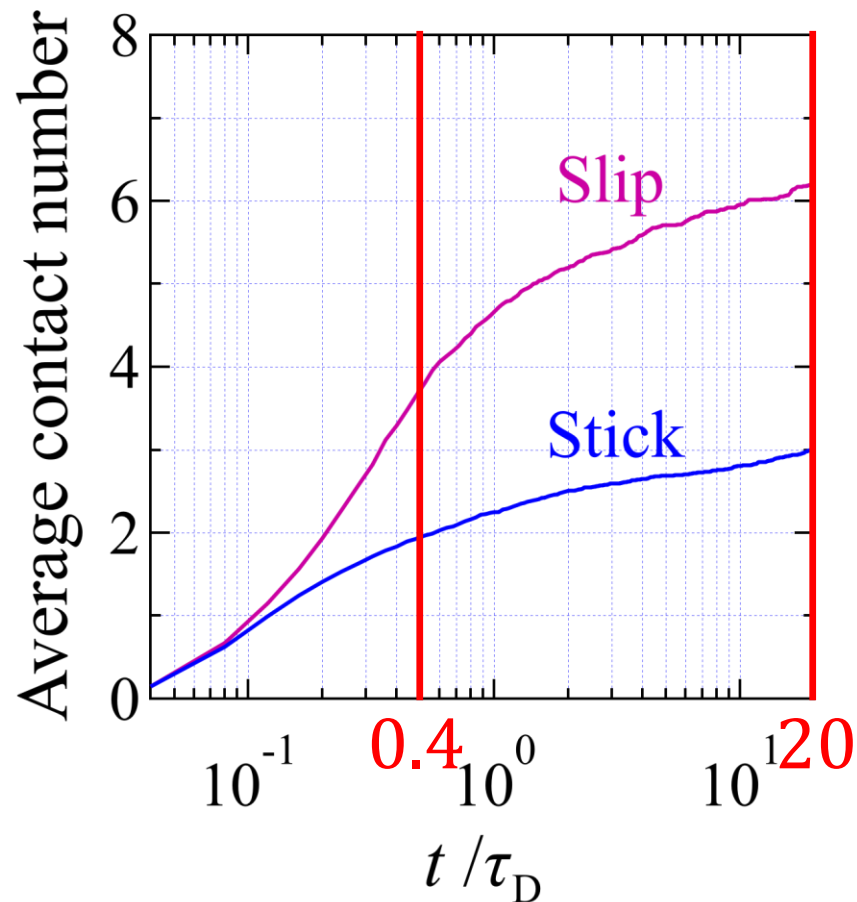


12



Contact number

0



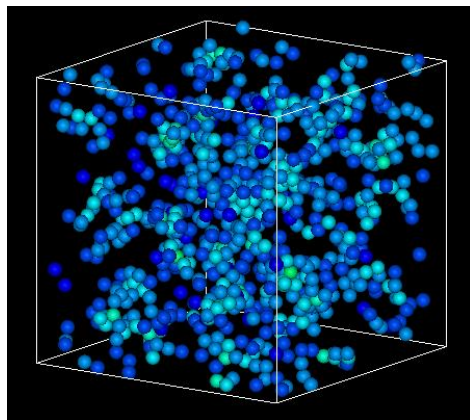
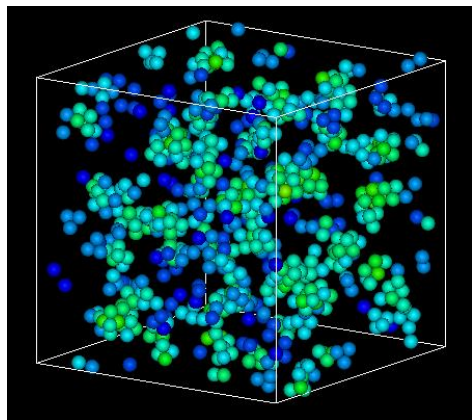
Characteristic time of
 particle diffusion: $\tau_D = d^2 / D$

Particle diameter: d Diffusion coefficient: D

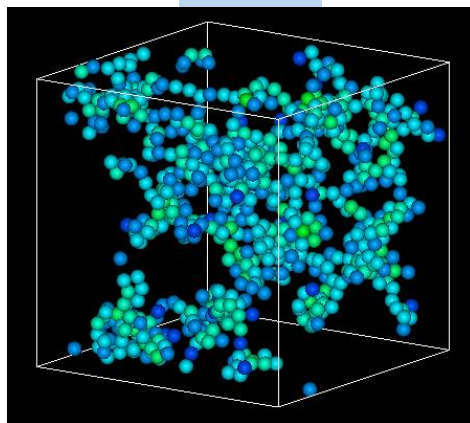
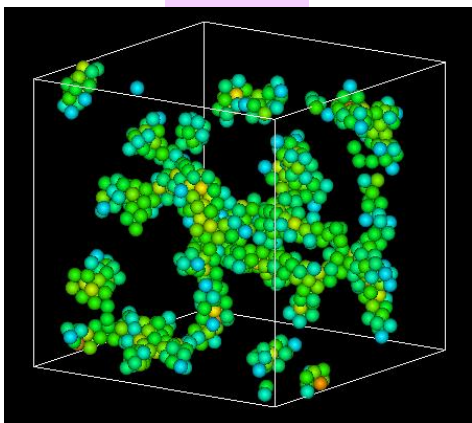
Aggregation

Slip

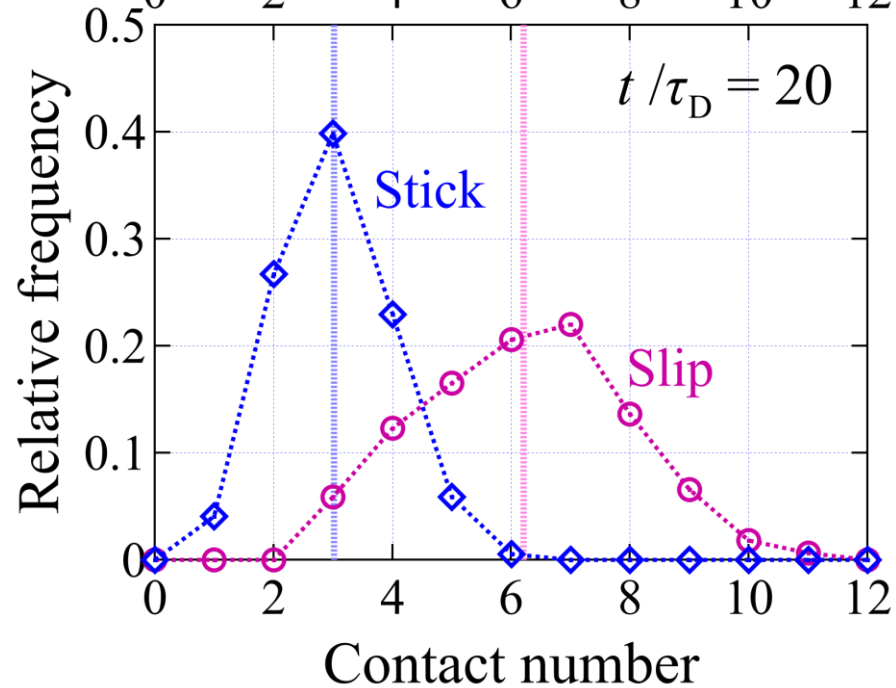
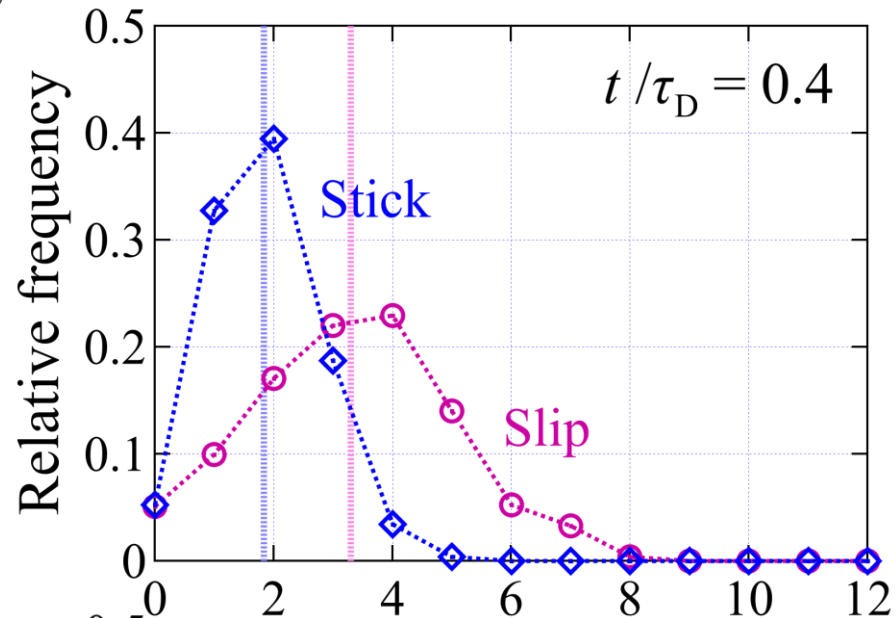
Stick



$t/\tau_D = 0.4$



$t/\tau_D = 20$



Simulation Conditions

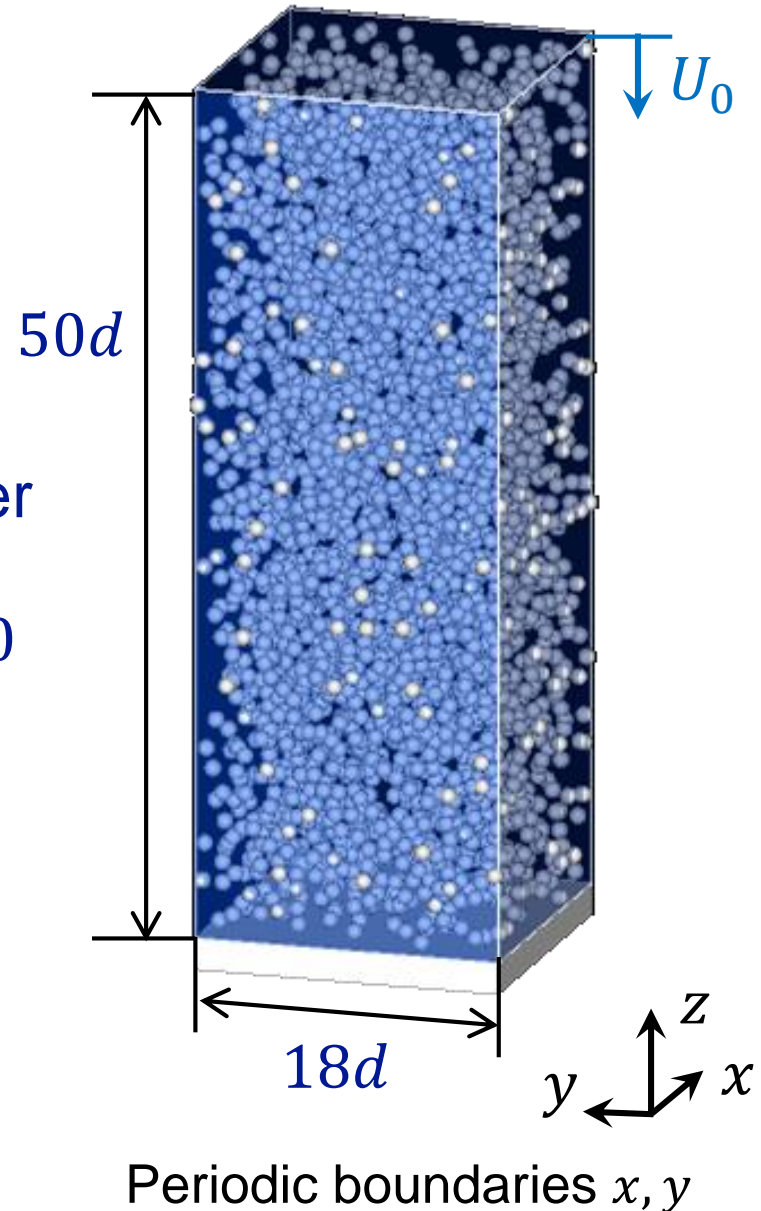
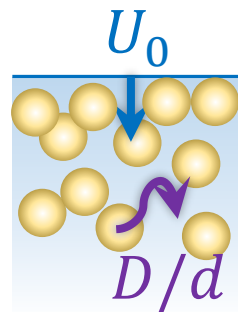
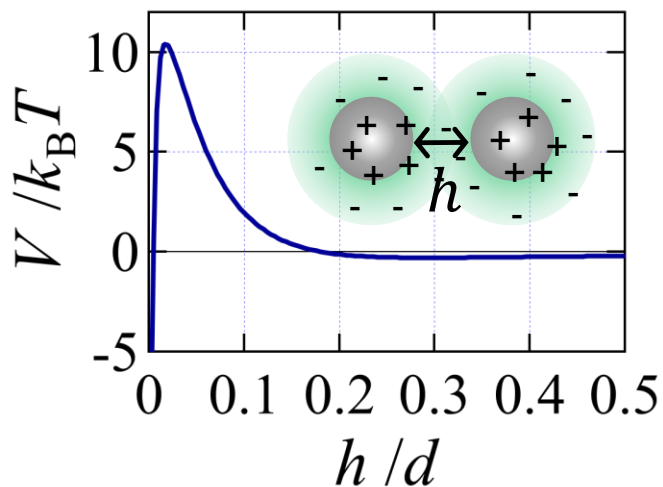
Particles

- Diameter $d = 20$ nm
- Initial volume fraction 10 vol%
- Zeta potential -50 mV
- Contact **Slip** / **Stick**

Liquid: water

- Initial particle drying Péclet number

$$Pe_0 = \frac{\text{(Drying rate)}}{\text{(Diffusion rate)}} = \frac{U_0}{D/d} = 400$$



Periodic boundaries x, y

Modeling of Falling Drying Rate

Drying rate:
$$\frac{U}{U_0} \approx \frac{R_0}{R_0 + R_p}$$

Resistance of evaporation: R_0

Resistance of particle layer: R_p



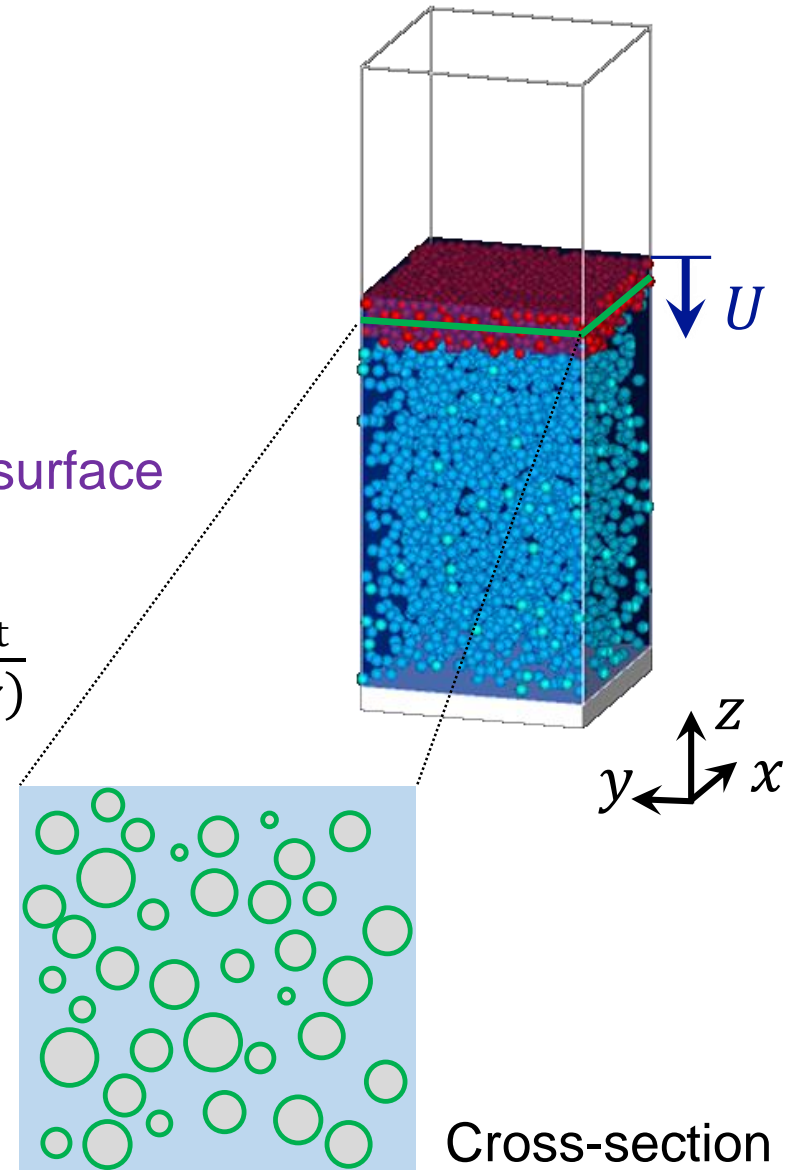
Aggregated particles moving with free surface

$$R_p = \int_P r(z) dz \quad r(z) = \frac{80}{[D_H(z)]^2} \frac{S_{\text{tot}}}{S_f(z)}$$

Hydraulic diameter: $D_H = \frac{4S_f}{L_f}$

Cross-sectional area of the flow: S_f

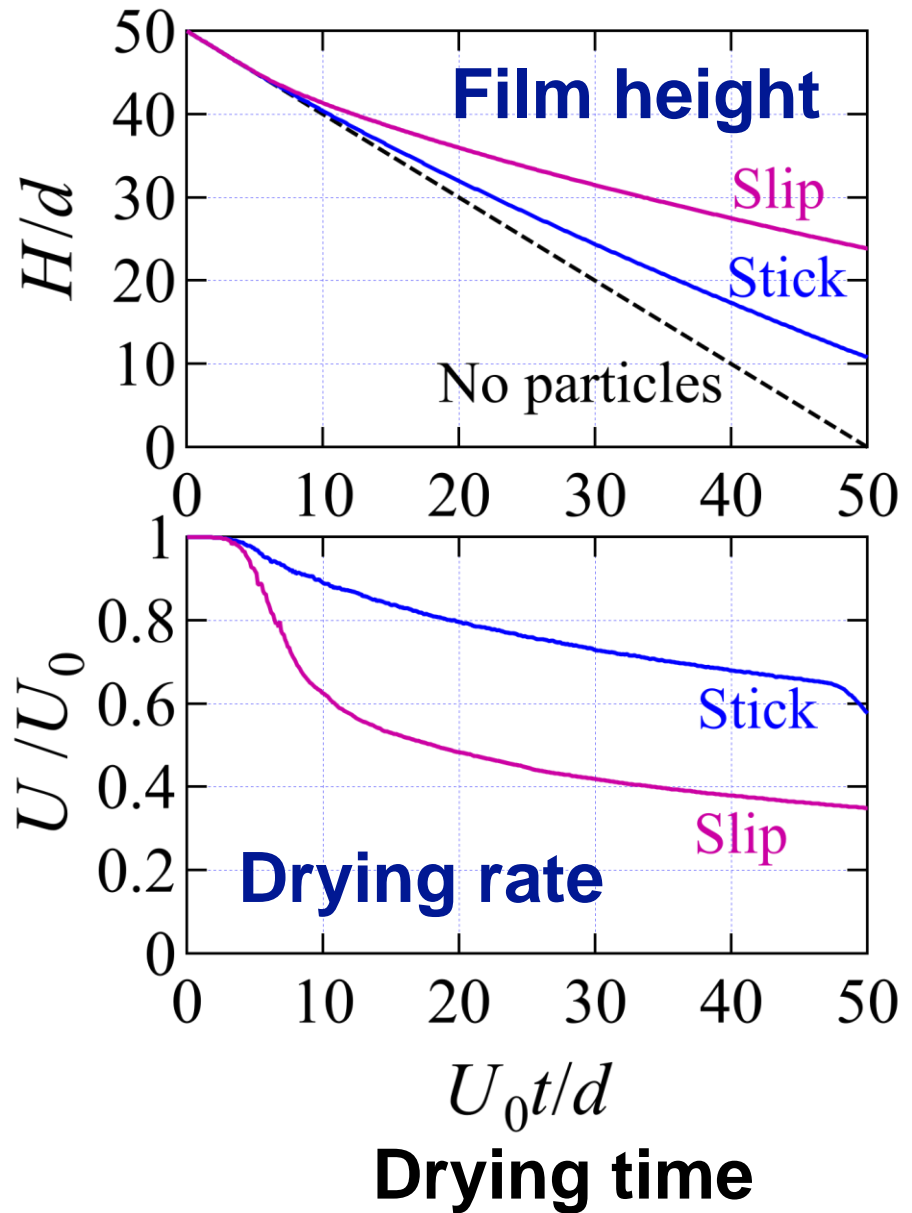
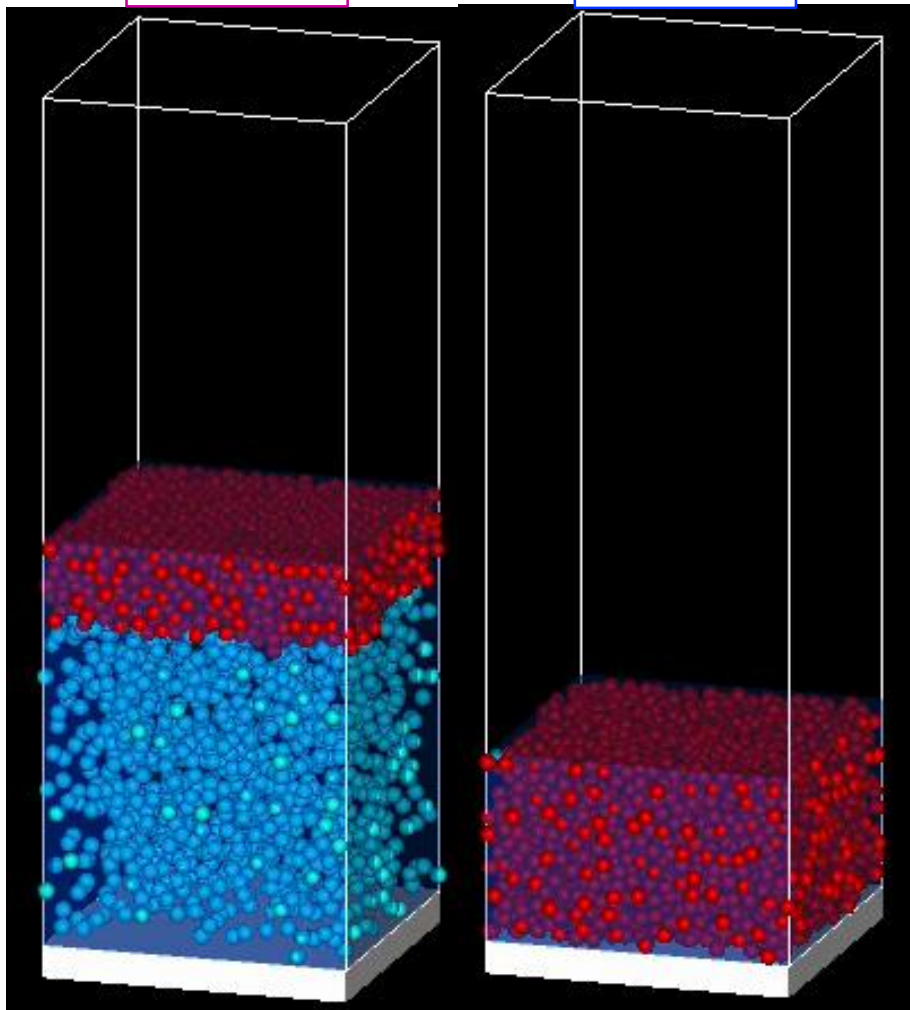
Wetted perimeter: L_f



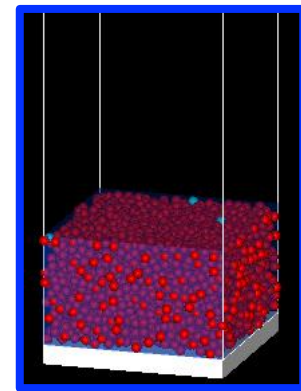
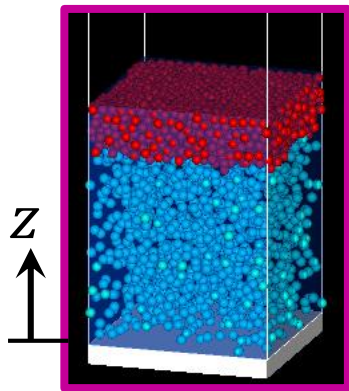
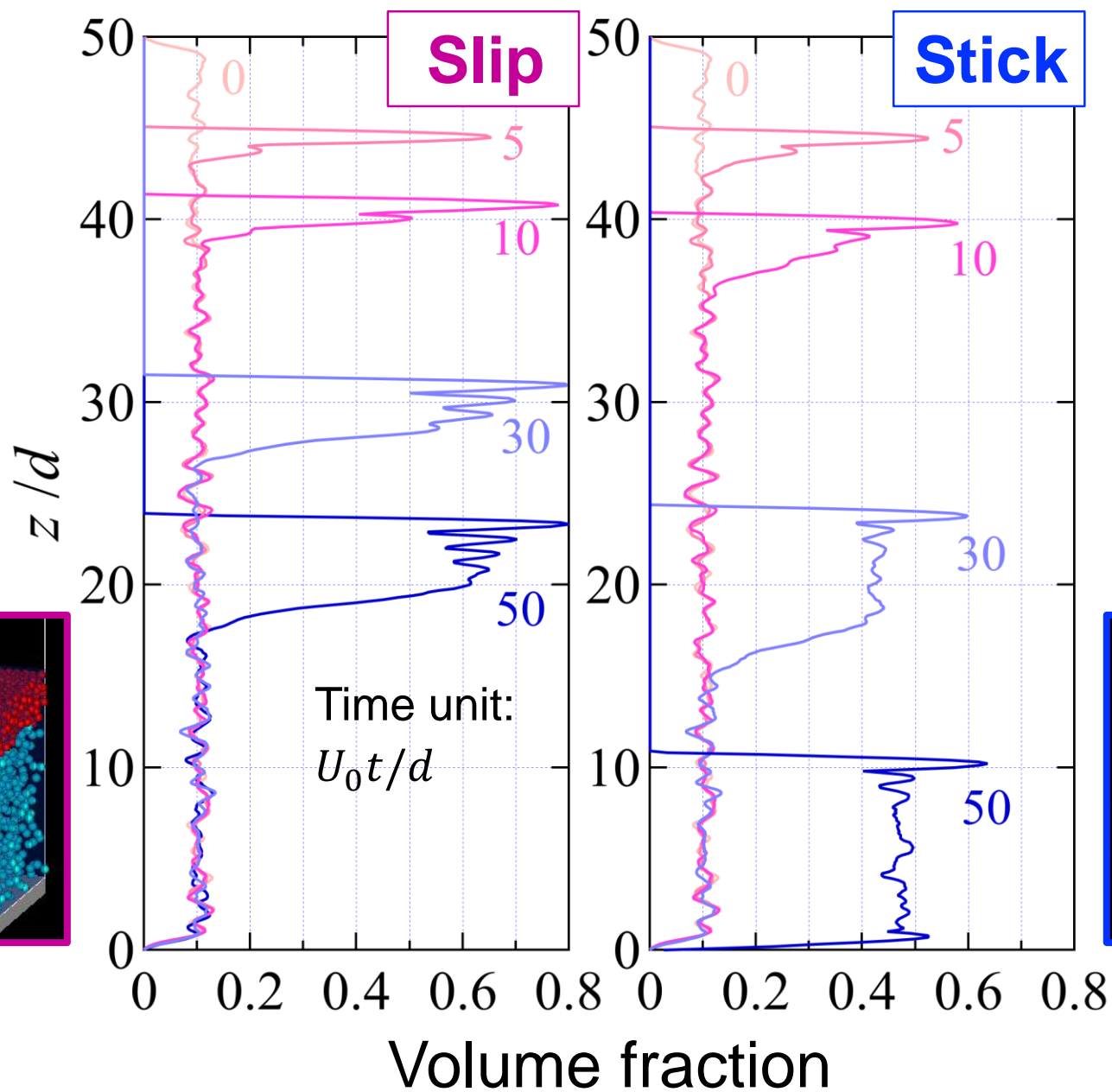
Drying Curves

Slip

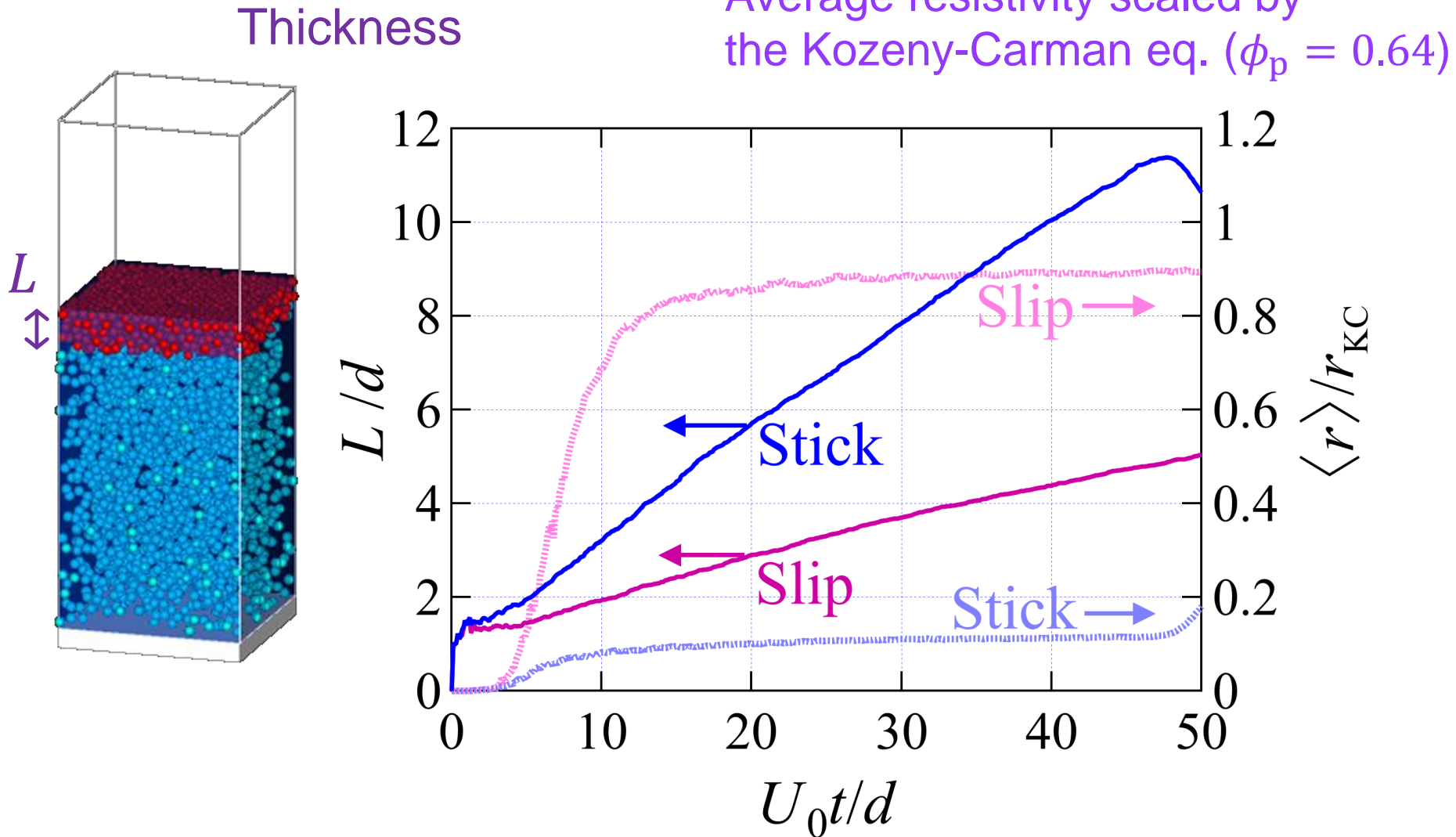
Stick



Particle Distribution



Growth of Particle Layer



Summary

- ◆ Modeling of adhesion between particles
 - Morphologies of aggregate
- Constraint on relative motions between contacting particles
 - Fixation of contact points
- Possible factor of adhesion in real systems: Binder addition
- ◆ Effects of adhesion on structure formation
 - Tree-like aggregates
 - Structures with high porosity formed during drying
 - High permeability (drying rate)